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# Covalent Bonding Workbook Answers Pearson Education

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8.1 Molecular Compounds > 38 Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved.  
•molecular compound: a compound that is composed of

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COVALENT BONDING Class Practice Problems In your notebook, solve the following problems. SECTION 8.1 MOLECULAR COMPOUNDS 1. Classify each of the following as an atom or a molecule. a. Be b. CO d. H<sub>2</sub>O e. Ne e. CO 2. Which of the following are diatomic molecules? a. CO<sub>2</sub> d. H<sub>2</sub>O 3. What types of elements tend to combine to form molecular compounds? 4.

The Octet Rule in Covalent Bonding Covalent compounds are most stable when each atom has eight electrons. Single, double, and triple covalent bonds depend on the number of pairs of electrons shared between two atoms. Atoms form double or triple covalent bonds if they can attain a noble gas structure by doing so. Type of Covalent Bond Single

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Covalent Bonds (pages 165–167) 1. Define a covalent bond. 2. A molecule is a group of atoms that are joined together by one or more covalent bonds. Circle the correct answer. negative neutral positive 3. Is the following sentence true or false? In a covalent bond, the atoms are held together by the attractions between the shared electrons and the

Bonding Basics - Covalent Bonds Answer Key/Teacher Notes Complete the chart for each element. Follow your teacher's directions to complete each covalent bond. (1) Hydrogen + Hydrogen (Diatomic Element) 1- Write the symbols for each element. 2 - Use Fruity Pebbles (or other cereal/candy with more

8.4 Polar Bonds and Molecules > 29 Copyright © Pearson Education, Inc., or its affiliates. All Rights Reserved. Hydrogen Bonds Attractions Between Molecules The